



## Catalytic Asymmetric Hydrosilylation of Ketones with New Chiral Ferrocenylphosphine-Imine Ligands

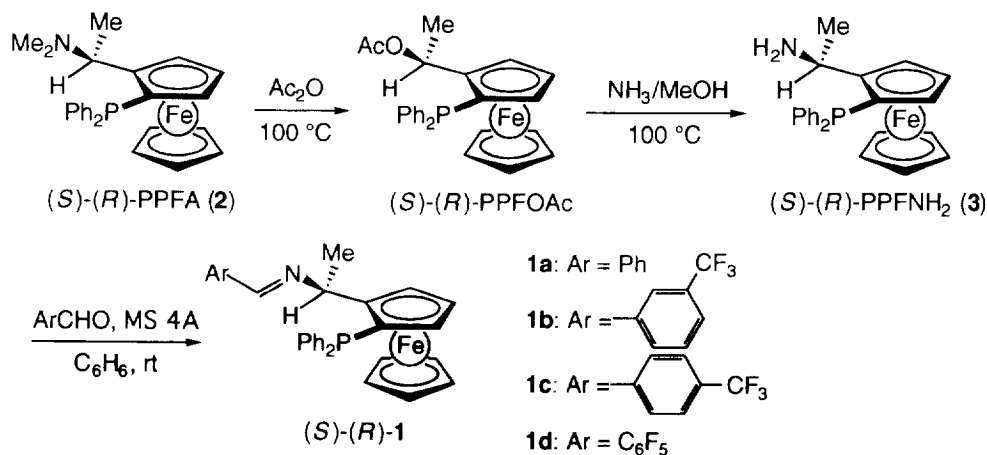
Tamio Hayashi,\* Chihiro Hayashi, and Yasuhiro Uozumi

Department of Chemistry, Faculty of Science, Kyoto University, Sakyo, Kyoto 606-01, Japan

**Abstract:** New chiral ferrocenylphosphines containing an imino group on the side chain, (*S*)-*N*-alkylidene-1-[(*R*)-2-diphenylphosphinoferoceanyl]ethylamines, were prepared by condensation of (*S*)-1-[(*R*)-2-diphenylphosphinoferoceanyl]ethylamine with aromatic aldehydes. The imino-phosphine ligands were found to be very effective for rhodium-catalyzed asymmetric hydrosilylation of prochiral ketones with diphenylsilane to give optically active alcohols of up to 90% ee.

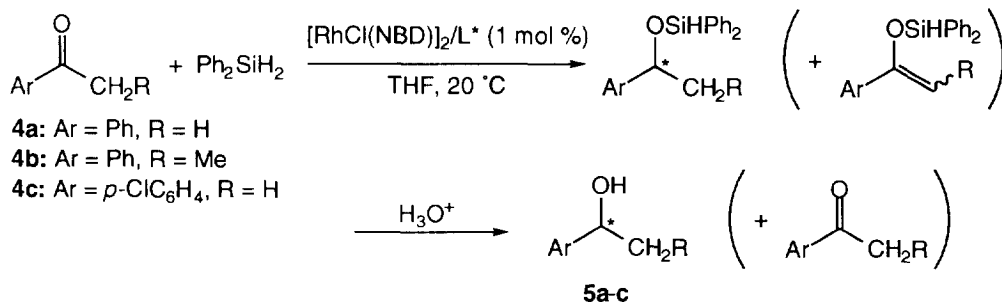
We have developed homochiral ferrocenylphosphines<sup>1</sup> which are effective as ligands for several types of asymmetric reactions catalyzed by transition metal complexes.<sup>2</sup> The ferrocenylphosphines have been demonstrated to be superior to others in that both monophosphines and 1,1'-bisphosphines can be readily prepared and structural modification can be readily made by introduction of a desired functional group on the side chain according to the demand of the reaction type. By this modification, the ferrocenylphosphines can bring about high enantioselectivity in a variety of catalytic asymmetric reactions, including rhodium(I)-catalyzed hydrogenation,<sup>3</sup> palladium(0)-catalyzed allylic substitution reactions,<sup>4,5</sup> and gold(I)- or silver(I)-catalyzed aldol-type reactions of isocyanocarboxylates.<sup>6,7,8</sup> Here we report that high enantioselectivity can be attained in the rhodium-catalyzed asymmetric hydrosilylation of ketones by use of new ferrocenylmonophosphines where an imino group was introduced on the ferrocenylmethyl position.

Scheme 1



A new ferrocenylmonophosphine, (*S*)-*N*-benzylidene-1-[(*R*)-2-diphenylphosphinoferrocenyl]ethylamine [(*S*)-(*R*)-**1a**] was prepared starting with (*S*)-(*R*)-PPFA **2** by a sequence of reactions shown in Scheme 1. The replacement of the dimethylamino group on (*S*)-(*R*)-PPFA **2** by an acetoxy group giving (*S*)-(*R*)-PPFOAc was previously reported to be effected by treatment with acetic anhydride at 100 °C.<sup>1</sup> The acetoxy group was substituted with an amino group by the reaction with a large excess of ammonia in methanol in an autoclave at 100 °C to give 80% yield of (*S*)-(*R*)-PPFNH<sub>2</sub> **3**.<sup>9</sup> The nucleophilic substitution reactions on the ferrocenylmethyl position are known to proceed with retention of configuration on the stereogenic carbon center.<sup>1</sup> Treatment of (*S*)-(*R*)-PPFNH<sub>2</sub> **3** with benzaldehyde in benzene in the presence of molecular sieves **4A** at room temperature gave 85% yield of imine (*S*)-(*R*)-**1a**, which contains diphenylphosphino group and an imino group on the same cyclopentadienyl ring. A chelate coordination to a metal is expected with the phosphorus atom and the imino nitrogen. The condensation of amine **3** with 3-trifluoromethylbenzaldehyde, 4-trifluoromethylbenzaldehyde, and pentafluorobenzaldehyde, in a similar manner gave the corresponding imino-phosphines, (*S*)-(*R*)-**1b**, **1c**, and **1d**, respectively.<sup>9</sup>

Scheme 2



The ferrocenylphosphine-imine ligands prepared here were examined for their enantioselectivity in the rhodium-catalyzed asymmetric hydrosilylation of prochiral ketones with diphenylsilane<sup>10</sup> (Scheme 2). The results are summarized in Table 1, which also contains data obtained with some other chiral ferrocenylphosphine ligands. The imino-phosphine (*S*)-(*R*)-**1a** was found to be an effective chiral ligand for the hydrosilylation of acetophenone. Thus, to a catalyst (1 mol %) solution generated by mixing (*S*)-(*R*)-**1a** with [RhCl(NBD)]<sub>2</sub> (P/Rh = 1.5/1.0) in THF was added successively acetophenone **4a** and diphenylsilane at 20 °C. It was found by a GLC analysis that all of the ketone was consumed in 1 h. Hydrolysis of the reaction mixture with dilute hydrochloric acid in methanol gave 90% yield of (*S*)-1-phenylethanol **5a**, whose enantiomeric purity was determined to be 87% ee by an HPLC analysis with a chiral stationary phase column (entry 1). The reaction of propiophenone **4b** also took place in the presence of (*S*)-(*R*)-**1a**-rhodium catalyst to give (*S*)-1-phenylpropanol **5b** of 86% ee (entry 2). A slightly higher enantioselectivity was observed in the reaction with the imino-phosphine ligands, **1b**, **1c**, and **1d**, which were derived from aldehydes containing electron-withdrawing groups on the phenyl ring. The hydrosilylation was faster with those ligands (entries 3-5). Thus, the reaction of **4a** was completed in 10 min with the imine ligand (*S*)-(*R*)-**1b**, which has 3-trifluoromethyl group on the phenyl, to give (*S*)-1-phenylethanol of 90% ee. The importance of imino group on the asymmetric hydrosilylation is demonstrated by the reaction with (*S*)-(*R*)-PPFA<sup>11</sup> **2** or (*S*)-(*R*)-PPFNH<sub>2</sub> **3** ligand, which gave 1-phenylethanol **5a** with opposite configuration (*R*) in much lower enantioselectivity (entries 7-8). It is

Table 1. Asymmetric Hydrosilylation of Ketones **4** with Ferrocenylphosphine-Rhodium Catalysts<sup>a</sup>

entry	ketone <b>4</b>	ligand	reaction <sup>b</sup> time	ratio <sup>c</sup> of <b>5/4</b>	yield (%) <sup>d</sup> of alcohol <b>5</b>	% ee <sup>e</sup> (config)
1	PhCOMe ( <b>4a</b> )	( <i>S</i> )-( <i>R</i> )- <b>1a</b> (Ar = Ph)	<1 h	>98/2	90	87 ( <i>S</i> )
2	PhCOEt ( <b>4b</b> )	( <i>S</i> )-( <i>R</i> )- <b>1a</b> (Ar = Ph)	<1 h	>98/2	97	86 ( <i>S</i> )
3	PhCOMe ( <b>4a</b> )	( <i>S</i> )-( <i>R</i> )- <b>1b</b> (Ar = <i>m</i> -CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> )	<10 min	>98/2	90	90 ( <i>S</i> )
4	PhCOMe ( <b>4a</b> )	( <i>S</i> )-( <i>R</i> )- <b>1c</b> (Ar = <i>p</i> -CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> )	<10 min	>98/2	94	89 ( <i>S</i> )
5	PhCOMe ( <b>4a</b> )	( <i>S</i> )-( <i>R</i> )- <b>1d</b> (Ar = C <sub>6</sub> F <sub>5</sub> )	<10 min	>97/3	86	89 ( <i>S</i> )
6	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub> COMe ( <b>4c</b> )	( <i>S</i> )-( <i>R</i> )- <b>1b</b> (Ar = <i>m</i> -CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> )	<1 h	>98/2	90	81 ( <i>S</i> )
7	PhCOMe ( <b>4a</b> )	( <i>S</i> )-( <i>R</i> )-PPFA ( <b>2</b> )	<1 h	74/26	64	16 ( <i>R</i> )
8	PhCOMe ( <b>4a</b> )	( <i>S</i> )-( <i>R</i> )-PPF-NH <sub>2</sub> ( <b>3</b> )	2 h	55/45	50	39 ( <i>R</i> )

<sup>a</sup> The reaction was carried out in THF in the presence of 1 mol % of the rhodium catalyst at 20 °C under nitrogen: THF (2.0 mL), ketone (2.0 mmol), diphenylsilane (2.5 mmol), [RhCl(NBD)]<sub>2</sub> (0.01 mmol), ligand (0.03 mmol). <sup>b</sup> The reaction time in which all of the ketone was consumed. <sup>c</sup> The ratio of alcohol/ketone determined by <sup>1</sup>H NMR of the products after hydrolysis. <sup>d</sup> Isolated yield by distillation. <sup>e</sup> Determined by HPLC analysis of (3,5-dinitrophenyl)carbamates of the alcohols with Sumichiral OA-4700 for **5a,c** and with Sumichiral OA-4100 for **5b** (*n*-hexane/1,2-dichloroethane/alcohol = 50/15/1).

interesting that formation of a considerable amount of silyl enol ether, which gives the starting ketone by the acidic hydrolysis, was observed with these ligands lacking the imino group.

As described above, the ferrocenylphosphine-imines **1** are effective as chiral ligands for the rhodium-catalyzed asymmetric hydrosilylation of ketones. Further investigation is in progress to design more stereoselective ferrocenylphosphine ligands.

#### Acknowledgment:

We thank the Ministry of Education, Japan, for a Grant-in-Aid for Scientific Research, for partial financial support of this work.

#### References and Notes:

- Hayashi, T.; Mise, T.; Fukushima, M.; Kagotani, M.; Nagashima, N.; Hamada, Y.; Matsumoto, A.; Kawakami, S.; Konishi, M.; Yamamoto, K.; Kumada, M. *Bull. Chem. Soc. Jpn.* **1980**, *53*, 1138.
- For reviews: (a) Hayashi, T. In *Ferrocenes*; Togni, A.; Hayashi, T., Eds.: VCH: Weinheim: **1995**, p. 105. (b) Sawamura, M.; Ito, Y. *Chem. Rev.* **1992**, *92*, 857. (c) Hayashi, T. *Pure Appl. Chem.* **1988**, *60*, 7. (d) Hayashi, T. In *Organic Synthesis: An Interdisciplinary Challenge*; Streith, J.; Prinzbach, H.; Schill G., Eds.: Blackwell Scientific Pub., Boston, **1985**, p. 35. (e) Hayashi, T.; Kumada, M. *Acc. Chem. Res.* **1982**, *15*, 395. (f) Hayashi, T.; Kumada, M. In *Fundamental Research in Homogeneous Catalysis*; Ishii, Y.; Tsutsui M., Eds.: Plenum Pub., New York, **1978**, Vol. 2; p. 159.
- (a) Hayashi, T.; Kawamura, N.; Ito, Y. *J. Am. Chem. Soc.* **1987**, *109*, 7876. (b) Hayashi, T.; Kawamura, N.; Ito, Y. *Tetrahedron Lett.* **1988**, *29*, 5969.
- (a) Hayashi, T.; Kishi, K.; Yamamoto, A.; Ito, Y. *Tetrahedron Lett.* **1990**, *31*, 1743. (b) Hayashi, T.; Yamamoto, A.; Ito, Y.; Nishioka, E.; Miura, H.; Yanagi, K. *J. Am. Chem. Soc.* **1989**, *111*, 6301. (c) Yamamoto, A.; Ito, Y.; Hayashi, T. *Tetrahedron Lett.* **1989**, *30*, 375. (d) Hayashi, T.; Kanehira, K.;

- Hagihara, T.; Kumada, M. *J. Org. Chem.* **1988**, *53*, 113. (e) Hayashi, T.; Yamamoto, A.; Ito, Y. *Tetrahedron Lett.* **1988**, *29*, 99. (f) Hayashi, T.; Yamamoto, A.; Ito, Y. *Chem. Lett.* **1987**, 177. (g) Hayashi, T.; Yamamoto, A.; Hagihara, T.; Ito, Y. *Tetrahedron Lett.* **1986**, *27*, 191. (h) Hayashi, T.; Yamamoto, A.; Ito, Y. *J. Chem. Soc., Chem. Commun.* **1986**, 1090.
- 5 Sawamura, M.; Nagata, H.; Sakamoto, H.; Ito, Y. *J. Am. Chem. Soc.* **1992**, *114*, 2586.
- 6 (a) Hayashi, T.; Sawamura, M.; Ito, Y. *Tetrahedron* **1992**, *48*, 1999-2012. (b) Ito, Y.; Sawamura, M.; Hamashima, H.; Emura, T.; Hayashi, T. *Tetrahedron Lett.* **1989**, *30*, 4681. (c) Sawamura, M.; Ito, Y.; Hayashi, T. *Tetrahedron Lett.* **1989**, *30*, 2247. (d) Ito, Y.; Sawamura, M.; Hayashi, T. *Tetrahedron Lett.* **1988**, *29*, 239. (e) Ito, Y.; Sawamura, M.; Kobayashi, M.; Hayashi, T. *Tetrahedron Lett.* **1988**, *29*, 6321. (f) Ito, Y.; Sawamura, M.; Shirakawa, E.; Hayashizaki, K.; Hayashi, T. *Tetrahedron Lett.* **1988**, *29*, 235. (g) Ito, Y.; Sawamura, M.; Shirakawa, E.; Hayashizaki, K.; Hayashi, T. *Tetrahedron* **1988**, *44*, 5253. (h) Ito, Y.; Sawamura, M.; Hayashi, T. *Tetrahedron Lett.* **1987**, *28*, 6215. (i) Ito, Y.; Sawamura, M.; Hayashi, T. *J. Am. Chem. Soc.* **1986**, *108*, 6405.
- 7 (a) Pastor, S. D.; Kesselring, R.; Togni, A. *J. Organomet. Chem.* **1992**, *429*, 415. (b) Pastor, S. D.; Togni, A. *Helv. Chim. Acta* **1991**, *74*, 905. (c) Togni, A.; Blumer, R. E.; Pregosin, P. S. *Helv. Chim. Acta* **1991**, *74*, 1533. (d) Togni, A.; Pastor, S. D. *J. Org. Chem.* **1990**, *55*, 1649. (e) Pastor, S. D.; Togni, A. *Tetrahedron Lett.* **1990**, *31*, 839. (f) Togni, A.; Pastor, S. D. *Tetrahedron Lett.* **1989**, *30*, 1071. (g) Togni, A.; Pastor, S. D.; Rihs, G. *Helv. Chim. Acta* **1989**, *72*, 1471. (h) Pastor, S. D.; Togni, A. *J. Am. Chem. Soc.* **1989**, *111*, 2333.
- 8 (a) Hayashi, T.; Uozumi, Y.; Yamazaki, A.; Sawamura, M.; Hamashima, H.; Ito, Y. *Tetrahedron Lett.* **1991**, *32*, 2799. (b) Hayashi, T.; Uozumi, Y.; Yamazaki, A.; Sawamura, M.; Hamashima, H.; Ito, Y. *Tetrahedron Lett.* **1991**, *32*, 2799. (c) Sawamura, M.; Hamashima, H.; Ito, Y. *J. Org. Chem.* **1990**, *55*, 5935.
- 9  $^1\text{H}$  (270 MHz,  $\text{CDCl}_3$ ) and  $^{31}\text{P}$  NMR (109 MHz,  $\text{CDCl}_3$ ) spectra and specific rotations for (*S*)-(*R*)-PPFNH<sub>2</sub> (**3**) and imino-phosphines (*S*)-(*R*)-**1** are as follows: **3**:  $^1\text{H}$  NMR  $\delta$  1.44 (d,  $J = 6.9$  Hz, 3 H), 1.44 (br s, 2 H), 3.77 (br s, 1 H), 4.02 (s, 5 H), 4.21 (dq,  $J = 2.3, 6.9$  Hz, 1 H), 4.28 (t,  $J = 2.3$  Hz, 1 H), 4.43 (br s, 1 H), 7.25-7.57 (m, 10 H).  $^{31}\text{P}$  NMR  $\delta$  -23.7.  $[\alpha]_{\text{D}}^{20} +296$  (c 0.51,  $\text{C}_6\text{H}_6$ ). **1a**:  $^1\text{H}$  NMR  $\delta$  1.65 (t,  $J = 6.6$  Hz, 3 H), 3.75 (br s, 1 H), 4.07 (s, 5 H), 4.33 (t,  $J = 2.3$  Hz, 1 H), 4.67 (br s, 1 H), 4.80 (dq,  $J = 2.3, 6.6$  Hz, 1 H), 6.73-7.54 (m, 15 H), 8.00 (s, 1 H).  $^{31}\text{P}$  NMR  $\delta$  -23.1.  $[\alpha]_{\text{D}}^{20} +387$  (c 0.50,  $\text{C}_6\text{H}_6$ ). **1b**:  $^1\text{H}$  NMR  $\delta$  1.65 (t,  $J = 6.6$  Hz, 3 H), 3.75 (br s, 1 H), 4.08 (s, 5 H), 4.35 (t,  $J = 2.3$  Hz, 1 H), 4.68 (br s, 1 H), 4.85 (dq,  $J = 2.3, 6.6$  Hz, 1 H), 6.67-7.57 (m, 14 H), 8.02 (s, 1 H).  $^{31}\text{P}$  NMR  $\delta$  -23.7.  $[\alpha]_{\text{D}}^{20} +372$  (c 0.50,  $\text{C}_6\text{H}_6$ ). **1c**:  $^1\text{H}$  NMR  $\delta$  1.68 (t,  $J = 6.6$  Hz, 3 H), 3.76 (br s, 1 H), 4.07 (s, 5 H), 4.34 (t,  $J = 2.3$  Hz, 1 H), 4.67 (br s, 1 H), 4.86 (dq,  $J = 2.3, 6.6$  Hz, 1 H), 6.70-7.52 (m, 14 H), 8.05 (s, 1 H).  $^{31}\text{P}$  NMR  $\delta$  -23.5.  $[\alpha]_{\text{D}}^{20} +352$  (c 0.50,  $\text{C}_6\text{H}_6$ ). **1d**:  $^1\text{H}$  NMR  $\delta$  1.69 (t,  $J = 6.6$  Hz, 3 H), 3.76 (br s, 1 H), 4.08 (s, 5 H), 4.35 (t,  $J = 2.6$  Hz, 1 H), 4.66 (m, 1 H), 4.82 (m, 1 H), 6.58-8.13 (m, 10 H), 10.0 (s, 1 H).  $^{31}\text{P}$  NMR  $\delta$  -23.8.  $[\alpha]_{\text{D}}^{20} +407$  (c 0.50,  $\text{C}_6\text{H}_6$ ).
- 10 For recent works on rhodium-catalyzed asymmetric hydrosilylation of ketones: (a) Nishiyama, H.; Kondo, M.; Nakamura, T.; Itoh, K. *Organometallics* **1991**, *10*, 500. (b) Sawamura, M.; Kuwano, H.; Ito, Y. *Angew. Chem. Int. Ed. Engl.* **1994**, *33*, 111. (c) Nishibayashi, Y.; Singh, J. D.; Segawa, K.; Fukuzawa, S.; Uemura, S. *J. Chem. Soc. Chem. Commun.* **1994**, 1375.
- 11 Hayashi, T.; Yamamoto, K.; Kumada, M. *Tetrahedron Lett.* **1974**, 4405.